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For pure substances (one-component systems), the chemical potential is the molar Gibbs energy;  $dG = (\mu_2 - \mu_1)dn$  so  $G = (\mu_2 - \mu_1)n = (-8.3 \text{ kJ mol}^{-1})(0.15 \times 10^{-3} \text{ mol}) = +1.2 \times 10^{-3} \text{ kJ} = 1.2 \text{ J}$ .

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