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Solutions 1) When dissolved

beryllium chloride reacts

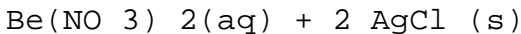
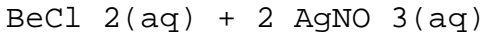
with dissolved silver

nitrate in water, aqueous

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beryllium nitrate and silver chloride powder are made.



Double Replacement 2) When

isopropanol ($\text{C}_3\text{H}_8\text{O}$)

burns in oxygen, carbon

dioxide, water, and heat are

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~~Zinc and Lead (II ...~~

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This is the most common type of chemical equation. In this chemical equation, a new product is formed by combining two to three combinations of reactants. For instance, $H_2 + O_2 \rightarrow H_2O$. This is a chemical

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equation where two atoms of hydrogen are combined to form a product, water. This is why this reaction is called as synthesis reaction.

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sodium chloride + lead (II)
nitrate lead (II) chloride +
sodium nitrate. 2. iron +
chlorine iron (III)

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chloride. 2. 3. barium +
water barium hydroxide +
hydrogen. 4. When chlorine
gas reacts with methane,
carbon tetrachloride and
hydrogen chloride are
produced. 5.

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~~Worksheet: Word Equations~~

Name

Word Equations Write the word equations below as chemical equations and balance: 1) Zinc and lead (II) nitrate react to form zinc nitrate and lead. 2)

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Aluminum bromide and chlorine gas react to form aluminum chloride and bromine gas. 3) Sodium phosphate and calcium chloride react to form calcium phosphate and sodium chloride.

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~~Balancing Equations~~

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Worksheet Balancing Word
Equations Chapter 10

(Remember the following are
diatomic: H₂, N₂, O₂, F₂,
Cl₂, Br₂, I₂) The

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coefficients should add up to the number at the end that is in parenthesis. 1.

carbon + oxygen carbon

dioxide (0) $C + O_2 \rightarrow CO_2$ 2.

copper + silver nitrate

copper(II) nitrate + silver

(4) $Cu + 2 AgNO_3 \rightarrow Cu(NO_3)_2$

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construct and deconstruct
the names of simple chemical
compounds.

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students and struggling scientists. The 'arrow' worksheet starts off gently and then moves on to unpicking more complicated information to create word equations.

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Equations from Chemical
Reactions~~

In each case complete the word equation by adding the name of the missing substance. (Explain your answers.) 1. sulfuric acid +

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sodium hydroxide ? _____ +

water I think this is the
answer because 2. iron +

_____ ? iron chloride +

copper I think this is the
answer because 3. _____ acid

+ magnesium carbonate ?

magnesium chloride + water +

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Carbon dioxide I think this
is the answer because

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(Age 7-11) 11+ (Age 7-11)

KS3 (Age 11-14) GCSE (Age 14-17) Spanish ESL Games Cup of Tea. Sugar ? carbon dioxide + ethanol. This is the word equation for fermentation, the process by which beer and wine are

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Services Program Write the equations for the following chemical reactions: 1) When dissolved barium chloride reacts with dissolved potassium sulfate in water, barium sulfate precipitate

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and aqueous potassium
chloride are made.

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Balancing an unbalanced
equation is mostly a matter
of making certain mass and

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charge are balanced on the reactants and products side of the reaction arrow. This is a collection of printable worksheets to practice balancing equations. The printable worksheets are provided in pdf format with

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