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\_\_\_ H<sub>3</sub>PO<sub>4</sub>(aq) + 2 H<sub>2</sub>O(l) + 3 H<sub>2</sub>PO<sub>4</sub><sup>-</sup>(aq)

Coefficient of H<sub>2</sub>SO<sub>4</sub> (aq) Coefficient of  
H<sub>3</sub>PO<sub>4</sub> (aq) A. 1 2 B. 2 3 C. 3 1 D. 3 2 3.

7.102 g of Na<sub>2</sub>SO<sub>4</sub> (142.04 g mol<sup>-1</sup>) is dissolved in water to prepare 0.5000 dm<sup>3</sup> of solution. What is the concentration of Na<sub>2</sub>SO<sub>4</sub> in mol dm<sup>-3</sup>? A. 2.500 × 10<sup>-2</sup> B. 1.000 × 10<sup>-1</sup> C. 1.000 ...

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