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question_answer Which of the following statement is false [BHU 1994] . A) The product of pressure and volume of fixed amount of a gas is independent of temperature . B) Molecules of different gases have the same K.E. at a given temperature

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(i) It is an inflammable gas (ii) It is the lightest gas. (iii) It is not easily liquefied. (iv) It is a strong oxidizing agent. (e) For the reaction PbO + H2 ? Pb + H 2 O, which of the following statements is wrong? i) H2 is the reducing agent. (ii) PbO is the oxidizing agent. (iii) PbO is oxidized to Pb (iv) H2 is oxidized to H 2 O.

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Study the following figure and answer questions. a) After heating Calcium carbonate, which gas is formed in a test tube? b) When we pass this gas through limewater what change, did you observe? c) Write down the chemical reaction showing the product formation after heating the Calcium carbonate.

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